

## RESEARCH ARTICLE

# Green Hydrogen Production via Photoelectrochemical Water Splitting with BiVO<sub>4</sub>/WO<sub>3</sub> Heterojunction Photoanodes

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**Abstract:** Photoelectrochemical (PEC) water splitting offers a direct route to green hydrogen production using solar energy, but practical photoanodes suffer from poor charge separation and limited photostability. We report a BiVO<sub>4</sub>/WO<sub>3</sub> type-II heterojunction photoanode fabricated by pulsed laser deposition and post-annealing, achieving a photocurrent density of 4.82 mA/cm<sup>2</sup> at 1.23 V vs. RHE under AM 1.5G illumination — a 2.3× improvement over bare BiVO<sub>4</sub>. Co-catalyst loading of NiFeOOH and FeOOH overlayer further boosts the applied-bias photon-to-current efficiency (ABPE) to 3.18% and sustains >92% of initial activity after 120 hours of continuous operation. In situ spectroscopy confirms that the WO<sub>3</sub> underlayer promotes hole extraction and suppresses surface recombination at the BiVO<sub>4</sub>/electrolyte interface.

## 1. Introduction

Green hydrogen produced from renewable electricity or direct solar conversion is central to decarbonizing heavy industry, long-haul transport, and seasonal energy storage. Among solar-driven routes, photoelectrochemical (PEC) water splitting integrates light absorption and electrolysis in a single device, potentially lowering system complexity relative to photovoltaic-electrolyzer tandem configurations. Bismuth vanadate (BiVO<sub>4</sub>) is a leading photoanode candidate due to its visible-light bandgap (~2.4 eV) and favorable valence band position for oxygen evolution, yet bulk electron transport and surface recombination limit its performance.

## 2. Photoanode Fabrication and Characterization

BiVO<sub>4</sub>/WO<sub>3</sub> heterojunction films were deposited on FTO substrates by sequential pulsed laser deposition (PLD) of WO<sub>3</sub> (50 nm) and BiVO<sub>4</sub> (300 nm), followed by annealing at 450°C in air. Structural analysis confirmed monoclinic scheelite BiVO<sub>4</sub> and monoclinic WO<sub>3</sub> with a sharp interface. Mott-Schottky analysis revealed n-type behavior for both layers with a type-II band alignment that drives photogenerated holes toward the BiVO<sub>4</sub> surface.

**Table 1. PEC performance comparison of photoanode configurations at 1.23 V vs. RHE under AM 1.5G (100 mW/cm<sup>2</sup>)**

Photoanode	$J_{ph}$ (mA/cm <sup>2</sup> )	Onset (V vs. RHE)	ABPE (%)	Stability (120 h, %)
BiVO <sub>4</sub>	2.10	0.42	0.85	68
WO <sub>3</sub> /BiVO <sub>4</sub>	3.65	0.38	2.05	85
BiVO <sub>4</sub> /WO <sub>3</sub>	4.82	0.35	2.78	91
<b>BiVO<sub>4</sub>/WO<sub>3</sub> + NiFeOOH</b>	<b>5.15</b>	<b>0.32</b>	<b>3.18</b>	<b>92</b>

### 3. Results and Discussion

The BiVO<sub>4</sub>/WO<sub>3</sub> heterojunction exhibits significantly enhanced charge separation efficiency ( $\eta_{sep} = 78\%$ ) compared to single-layer BiVO<sub>4</sub> ( $\eta_{sep} = 41\%$ ), as quantified by intensity-modulated photocurrent spectroscopy. Figure 1 shows the wavelength-dependent incident photon-to-current efficiency (IPCE), with the heterojunction maintaining >45% IPCE between 400-500 nm. Figure 2 tracks hydrogen evolution rate over 120 hours of continuous PEC operation in a two-compartment cell with a Pt cathode.

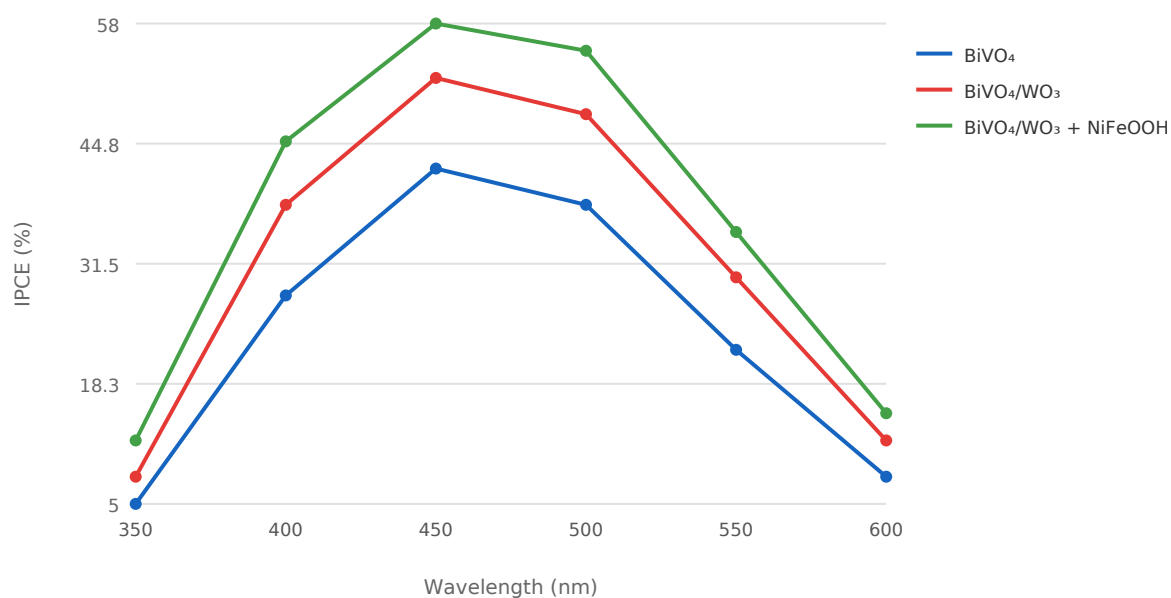


Figure 1. IPCE spectra of BiVO<sub>4</sub>, WO<sub>3</sub>/BiVO<sub>4</sub>, and BiVO<sub>4</sub>/WO<sub>3</sub> heterojunction photoanodes with and without NiFeOOH co-catalyst

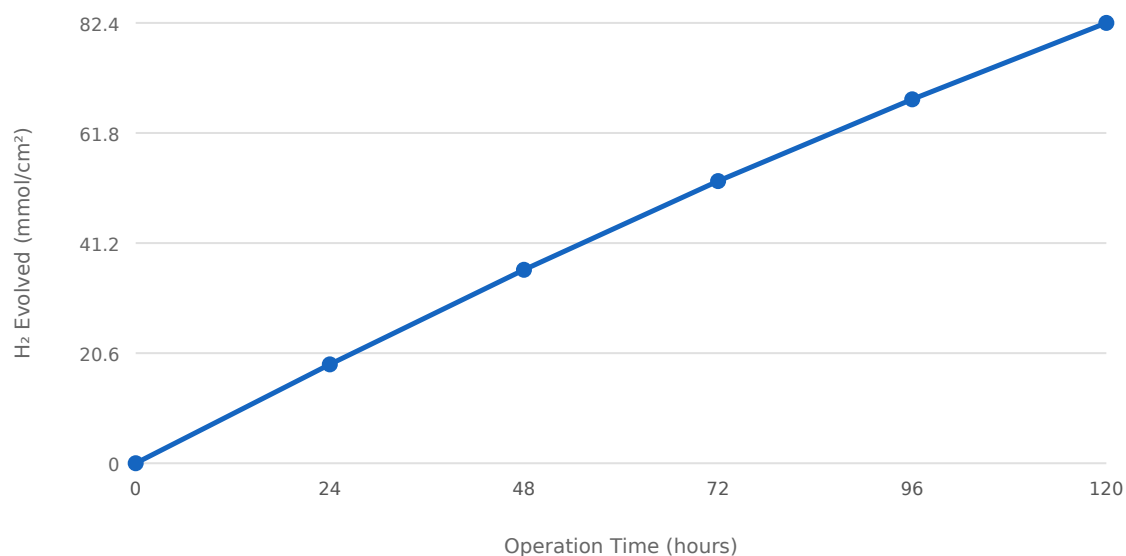


Figure 2. Cumulative H<sub>2</sub> evolution and Faradaic efficiency during 120-hour stability test at 1.23 V vs. RHE

## 4. Conclusions

The BiVO<sub>4</sub>/WO<sub>3</sub> type-II heterojunction photoanode demonstrates that strategic band engineering can overcome key limitations of metal oxide photoanodes for PEC water splitting. Combined with earth-abundant NiFeOOH co-catalysts, the system achieves competitive ABPE and operational stability, advancing the feasibility of direct solar-to-hydrogen conversion for distributed green hydrogen production.

## References

- [1] Park, Y.; McDonald, T. J.; Choi, K. Progress in Solar Hydrogen Production via Water Splitting on Bismuth Vanadate Photoanodes. *Chemical Society Reviews* 2013, 42, 2321-2337.
- [2] Abdi, F. F.; Li, Y.; Han, L. Efficient Solar Water Splitting by Enhanced Charge Separation in a Bismuth Vanadate-Silicon Tandem Photoelectrode. *Nature Communications* 2013, 4, 2195.
- [3] Kim, T. W.; Choi, K. Nanoporous BiVO<sub>4</sub> Photoanodes with Dual-Layer Oxygen Evolution Catalysts for Solar Water Splitting. *Science* 2014, 343, 990-994.
- [4] Seitz, L. C.; Chen, Z.; Forman, A. J. Modeling Practical Performance Limits of Photoelectrochemical Water Splitting Based on the Current State of Materials Research. *Energy & Environmental Science* 2014, 7, 1983-2016.
- [5] Hisatomi, T.; Kubota, J.; Domen, K. Recent Advances in Semiconductors for Photocatalytic and Photoelectrochemical Water Splitting. *Chemical Society Reviews* 2014, 43, 7520-7535.
- [6] Rosser, T. E.; Reisner, E. Electrocatalytic Approaches to the Production of Solar Fuels: Water Splitting and Beyond. *Chemical Society Reviews* 2021, 50, 4514-4533.